

CHEMISTRY OF BLEOMYCIN. XVI
 EPI-BLEOMYCIN

Sir:

As previously reported^{1,2}, migration of the carbamoyl group from the 3- to the 2-hydroxyl group of the mannose moiety of bleomycin occurs in aqueous alcoholic solution of copper-free bleomycin in the presence of a weak base such as triethylamine until an equilibrium is established. This carbamoyl-migration product has been called *iso*-bleomycin. Under the same reaction condition, copper-chelated bleomycin does not give *iso*-bleomycin³, but yields irreversibly another product designated *epi*-bleomycin. In this communication, the formation, isolation, chemical and biological properties and structure of *epi*-bleomycin are reported.

Preliminary experiments showed that epimerization proceeds slower than isomerization. In order to study the reaction process, liquid chromatography, as shown in Fig. 1, was used for the rapid quantitative determination of the reaction product. This analytical method is also useful for determination of the products derived by chemical and biological reactions in general, although minor modifications of the solvent system are necessary in some case.

To 1 ml of a 10% aqueous solution of copper-chelated bleomycin B₂ was added 2 ml of ethanol containing 2% triethylamine. The reaction mixture was divided into two fractions, which were kept at 25°C and 70°C, and analyzed intermittently by liquid chromatography. At both temperatures, the reaction apparently proceeded with first order. Fifty percent conversion was achieved in about 120 hours at 25°C, and in about 1.5 hour at 70°C. At 70°C in 6 hours, more than 90% of bleomycin B₂ was converted to copper-chelated *epi*-bleomycin B₂. Copper-chelated *epi*-bleomycin B₂ was preparatively isolated by CM-Sephadex C-25 column chromatography using 0.15 M sodium acetate-acetic acid buffer at pH 4.5. *Epi*-bleomycin B₂ was eluted faster than the intact bleomycin. Under these conditions *iso*-bleomycin is eluted later than bleomycin.³ Other *epi*-bleomycins showed similar chromatographic behavior. Thus, *epi*-bleomycins A₂, B₂, A₂'-C, and B₄ were isolated in the pure state.

¹³C-NMR studies of copper-free bleomycin and *epi*-bleomycin, which were obtained by treatment

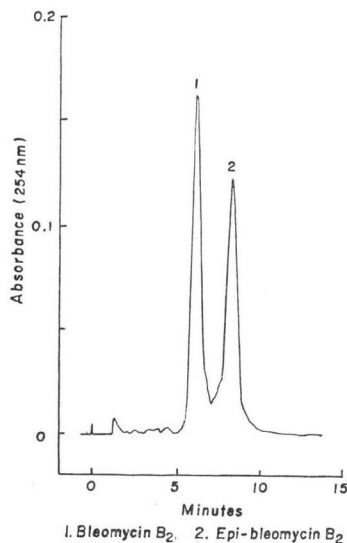
of copper-chelated materials with hydrogen sulfide, showed that they have the same number of carbon atoms. The results of the N- and C-terminal analyses suggested that no bond cleavage occurred during epimerization.

To test whether carbamoyl migration occurred during epimerization as in the case of *iso*-bleomycin or not, copper-free *epi*-bleomycin was methanolized using Amberlyst 15 as an acid catalyst, and the resulting methyl glycosides of the sugar components were analyzed by thin-layer chromatography after O-benzoylation^{1,2}. The result indicated that gulose and 3-O-carbamoyl-mannose exist in *epi*-bleomycin molecule, but 2-O-carbamoyl-mannose does not.

Copper-chelated *epi*-bleomycin showed essentially the same UV spectrum as copper-chelated bleomycin. The CD spectra were distinctly different (Fig. 2), which suggested conversion of an asymmetric carbon atom adjacent to a chromophore.

Fig. 1. Separation of bleomycin B₂ and *epi*-bleomycin B₂ by liquid chromatography

Sample:	Reaction mixture (70°C, 1.5 hours) 33 μg (see text)
Instrument:	Perkin-Elmer 1220
Column:	LiChrosorb SI 60 (5 μ)
Dimensions:	150 × 2.6 mm (ID)
Mobile Phase:	10% AcONH ₄ -H ₂ O-MeOH-EtOH (5: 19: 60: 116)
Flow Rate:	0.5 ml/min
Temperature:	60°C
Pressure:	600 psi
Detector:	UV (254 nm)



The amine components obtained by total acid hydrolysis of copper-free *epi*-bleomycin A₂ were analyzed by cellulose thin-layer chromatography, cellulose electrophoresis and in an automatic amino acid analyzer. All amine components (I~VII) present in the total acid hydrolyzate of bleomycin A₂³⁾ (Table 1) were found in the hydrolyzate of *epi*-bleomycin A₂. The diastereomers of components I, III⁴⁾, and IV⁵⁾ (Table 1) were not detected by automatic amino acid analysis. This means that the asymmetric carbons of these three amine components kept their configuration intact during epimerization.

As already reported component II (Table 1) is partially racemized during total acid hydrolysis⁶⁾. The racemate of II can be easily resolved by cellulose thin-layer chromatography using 1-propanol - pyridine - acetic acid-water (15: 10: 3: 12)⁶⁾. The thin-layer chromatograms of the hydrolyzates of bleomycin and *epi*-bleomycin were analyzed by a double-wavelength UV densitometry⁷⁾, as shown in Fig. 3. The R_f-values of the intact II (S-configuration⁸⁾, designated S-II) and its enantiomer (designated R-II) are 0.11 and 0.20, respectively. As shown in Fig. 3, in the hydrolyzate of bleomycin the peak of S-II is higher than that of R-II, while in the case of *epi*-bleomycin the peak heights are reversed. The results indicate that the configuration of the methine carbon of the β-lactam ring of *epi*-bleomycin is converted to R-configuration during epimerization.

A simple method for determination of R- or S-V (Table 1) has not been established. Therefore, V was isolated from the hydrolyzate of *epi*-bleomycin A₂ by ion-exchange column chromatography⁹⁾. The optical rotatory dispersion spectrum indicated that it has S-configuration, although it was partially racemized during acid hydrolysis*.

From the above experimental results, it can be concluded that bleomycin and *epi*-bleomycin are different only in the configuration of the methine carbon of the β-lactam ring. The reason why copper-chelated bleomycin is epimerized in basic

Fig. 2. CD Spectra of copper-chelated *epi*-bleomycin A₂ and copper-chelated bleomycin A₂

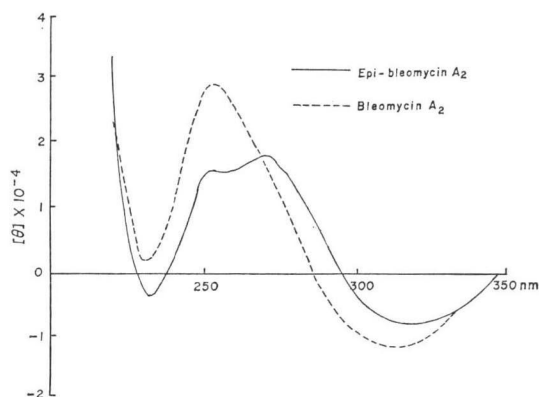


Table 1. Amine components of *epi*-bleomycin A₂ and bleomycin A₂.

Amine component	Number of asymmetric carbon	Stereochemistry	
		<i>Epi</i> -bleomycin A ₂	Bleomycin A ₂
I <chem>CC(O)C(N)C(=O)O</chem>	2	2S, 3R	2S, 3R ³⁾
II <chem>CC1=C(N)NC(=O)C=C1C(=O)O</chem>	1	R*	S* ⁷⁾
III <chem>CC(O)C(N)C(C)C(=O)O</chem>	3	2S, 3S, 4R	2S, 3S, 4R ¹⁰⁾
IV <chem>CC1=CN=C(N)C=C1C(O)C(N)C(=O)O</chem>	2	2S, 3R	2S, 3R ¹¹⁾
V <chem>NC(C)C(N)C(=O)O</chem>	1	S*	S* ⁸⁾
VI <chem>NC(C)C(N)C1=NC=NC=C1C(=O)O</chem>	0		
VII <chem>NC(C)C(N)C(C)C(=O)O</chem>	0		

* Partially racemized during acid hydrolysis

aqueous alcoholic solution while copper-free bleomycin is isomerized under the same conditions may be as follows. A nitrogen atom of the pyrimidine chromophore is one of the copper-ligands of bleomycin, as verified by the UV spectrum. Complex formation causes the activation of the adjacent chiral methine, which is, therefore, easily epimerized with base.

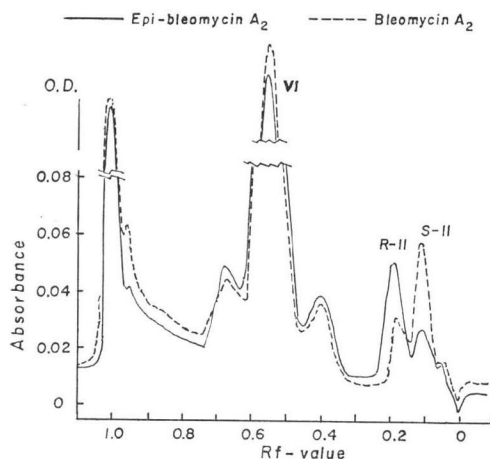
We should add a further brief comment on the copper-ligands of the bleomycin molecule. One

* Partial racemization was also observed with bleomycin A₂.

Fig. 3. Densitogram of TLC of hydrolyzates of *epi*-bleomycin A₂ and bleomycin A₂

Precoated Avicel SF, 1-propanol - pyridine - acetic acid - water (15: 10: 3: 12).

Detection: Hitachi 356 double-wavelength spectrophotometer with TLC accessories, $\lambda_1=350$ nm, $\lambda_2=290$ nm.



ligand is the nitrogen atom of the imidazole of IV, as can be deduced from the fact that the pKa value of 4.7 due to the imidazole of copper-free bleomycin (B₂) disappears in the copper-complex and copper-free bleomycin gives positive PAULI reaction but copper-complex does not. Another ligand is the α -amino group of V. The reactivity of this amino function is markedly reduced by copper-chelate formation⁹⁾ and the pKa value 7.3 due to this amino group of the copper-free bleomycin disappears in the copper-complex. The other ligand would be the carbamoyl group, as suggested by lack of migration of this group in the copper-complex.

The antimicrobial activity of *epi*-bleomycins was compared with that of the intact bleomycins (Table 2). The antimicrobial activity against *Mycobacterium* 607 of *epi*-bleomycins was about 10~20% that of the original ones. But the anti-*B. subtilis* activity was much less than the anti-*Mycobacterium* activity. Concentrations of *epi*-bleomycin A₂ and B₂ required for fifty percent growth inhibition of HeLa cell (ID₅₀) were 9.8 and 8.8 μ g/ml, while the ID₅₀s of the normal compounds were 2.8 and 2.2 μ g/ml, respectively.

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Table 2. Antimicrobial activity of *epi*-bleomycins and intact bleomycins.

Bleomycin	<i>Mycobacterium</i> 607	<i>Bacillus subtilis</i>
<i>epi</i> -A ₂ (+Cu)	93 u/mg*	3 u/mg*
A ₂ (+Cu)	910	910
<i>epi</i> -B ₂ (+Cu)	584	15
B ₂ (+Cu)	2,720	1,460
<i>epi</i> -A ₂ '-C(+Cu)	69	trace
A ₂ '-C (+Cu)	847	548
<i>epi</i> -B ₄ (+Cu)	675	175
B ₄ (+Cu)	6,350	14,500

* Cup-assay method, bleomycin A₂ (-Cu) free base as the standard: 1,000 u/mg.

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